



# Green Processes for low cost and high productivity

绿色制程带来低成本与高生产力

Solarcon China. March 2010

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# Fluorine Environmental Benefit

## 氟气对环境的好处

## What do we mean by “Green”? 您的产品被称为 « 绿色 » ?



Is your product “Green” only because it uses less energy?

Is your process “Green” because you can’t detect any emissions from your factory?

An electric car is only as clean as the power station that generates it’s power!



Clean?



Green?

The full supply chain and product lifecycle must be considered: Energy use, environmental impact for all elements of the supply chain.

The same is true for Green House Gases.....

## Trend to regulate Green House Gas Emissions from silicon processing

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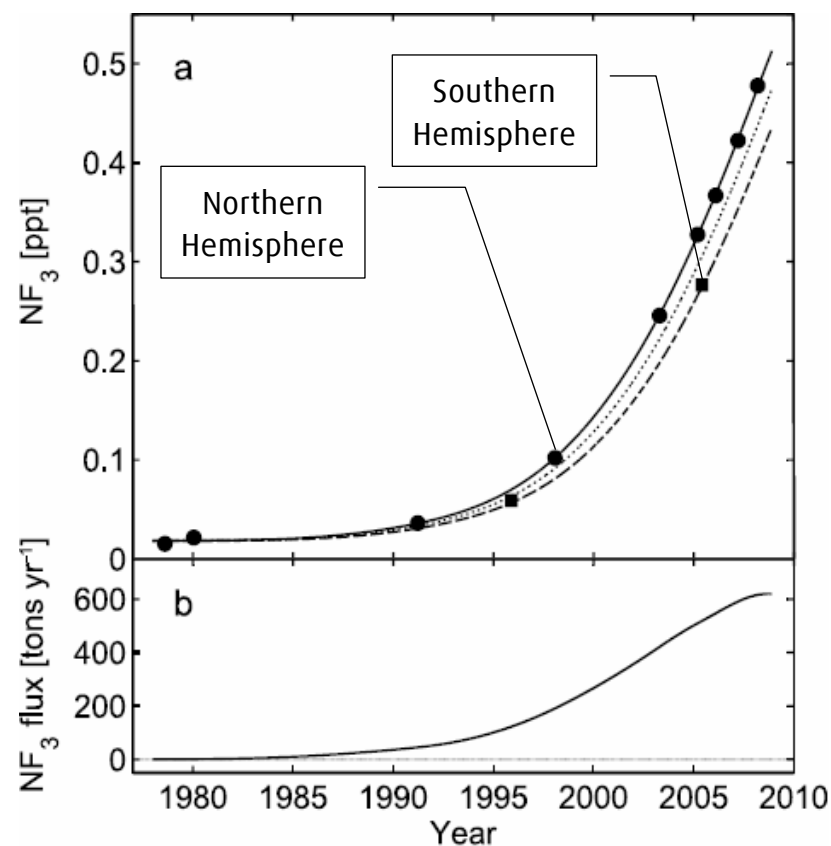
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June 2008: Michael Prather identifies  $\text{NF}_3$  as “The Greenhouse Gas Missing from Kyoto”  
«  $\text{NF}_3$ , 京都议定书中遗漏的温室气体 »

October 2008: Scripps Institute measure  $\text{NF}_3$  in the atmosphere,  
~16% of global capacity is vented!

April 2009: US EPA legislation requires use and emissions monitoring for every process using GHGs.

UNIPCC –  $\text{NF}_3$  is expected to be listed as a GHG.



**Legislation usually is developed to address OVERALL ACTUAL measured emissions of the material, from the entire industry, not just the point of use!**

# Semiconductor Industry Driving Forces

## ITRS ESH 2007 – Risk of restriction of Chemical use

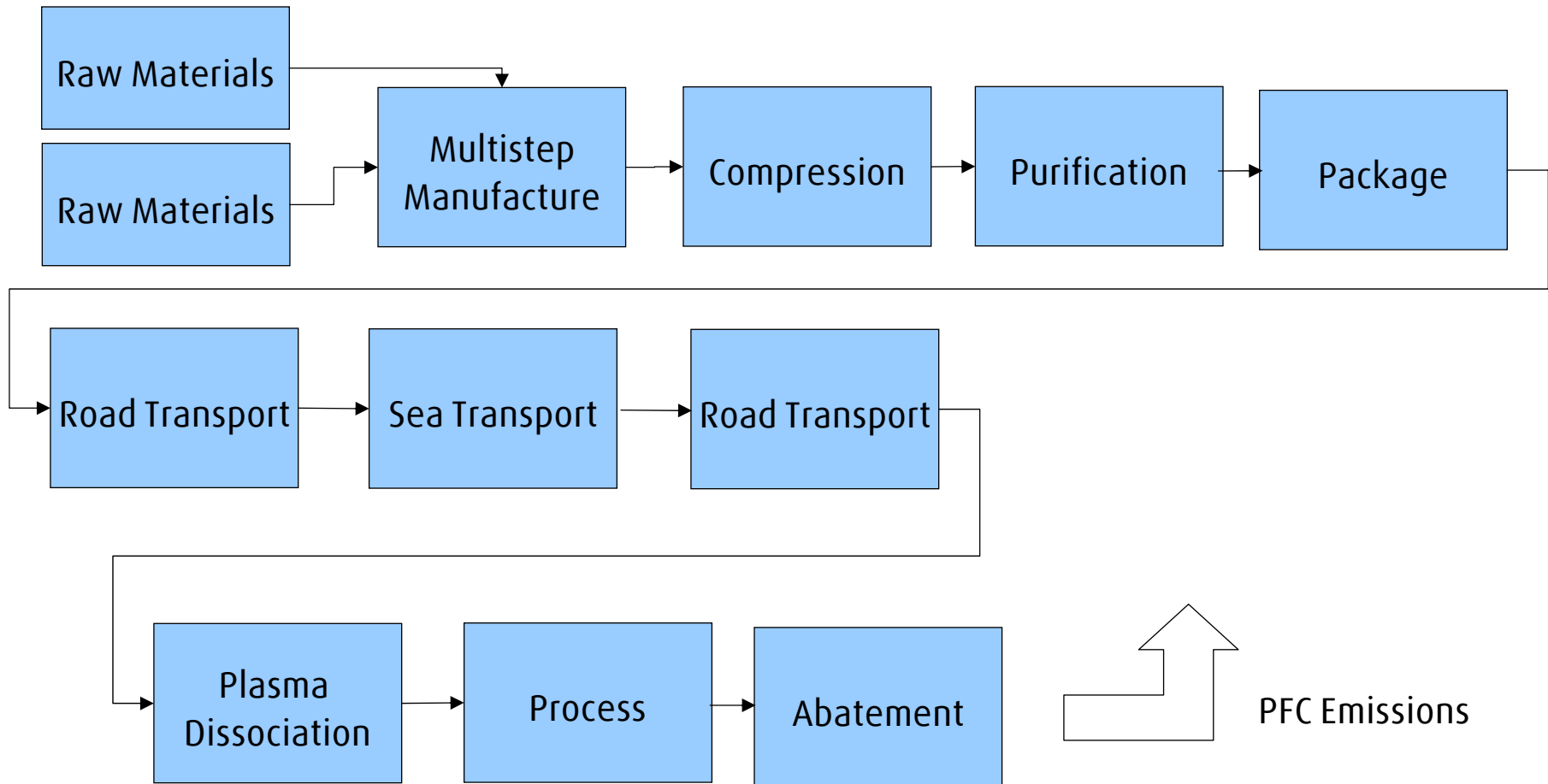


Issues & Characterization	Show Stopper	High Restriction Potential	Medium Restriction Potential
<b>List of Chemicals or Raw Materials Subject to Actual or Potential Manufacture or Use Restrictions</b>	Asbestos Materials Certain glycol ethers Polychlorinated biphenyls Fully halogenated chlorofluorocarbons (CFCs) Carbon tetrachloride 1,1,1 trichloroethane Halons 1211, 1301, 2402 Hydrobromofluorocarbons (HBFCs) HCFC 141b Polybrominated biphenyls (PBBs) and their ethers/oxides (PBDEs) Cadmium compounds Lead compounds Mercury compounds Hexavalent Chromium compounds Polychlorinated Biphenyls (PCB)/ Terphenyls (PCT) Polychlorinated Naphthalene (PCN) Short Chain Chlorinated Paraffins (C10-13, Cl >50%) Tributyl tin (TBT) and, Triphenyl tin (TPT) compounds Certain Azo Colorants	Hydrochlorofluorocarbons (HCFCs) Perfluorooctyl sulfonates (PFOS) Cadmium compounds Lead compounds Mercury compounds Hexavalent Chromium compounds Other chlorinated organic compounds Other brominated organic compounds	Perfluorocompounds (PFCs) - SF6                      - C4F10 - C2F6                    - C5F12 - CF4                      - C6F14 - NF3 - C4F8 - CHF3 - C3F8 Hydrofluorocarbons (HFCs) Perfluorooctanoic acid (PFOA) and its salts Certain phthalates Phenols Perfluoroalkyl sulfonates (PFAS) Ethylene Oxide Ethylene Dichloride Polyaromatic hydrocarbons Antimony Trioxide Beryllium Polyvinyl chloride (PVC) Other brominated flame retardants

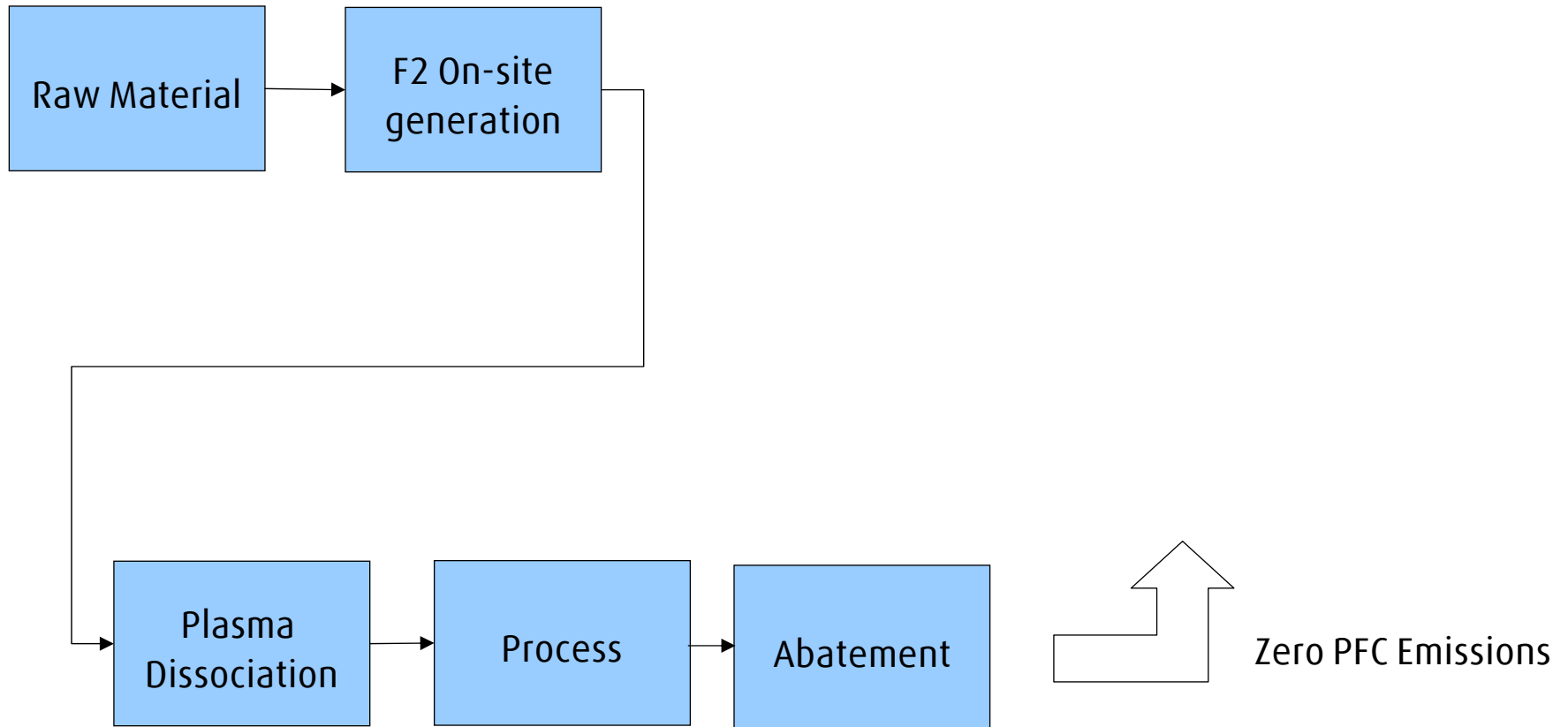
**Many typical cleaning gases may become restricted**



# NF<sub>3</sub> : Multistep Manufacture – Energy intensive



**F<sub>2</sub> On-site Supply: Simpler process.  
More efficient, less capital intensive**





# CO2 equivalent savings from elimination of NF3



NF3 -> F2 larger scale customers: Malibu, Masdar, TMD and LG Display

Annual CO2 emissions eliminated from the cleaning gas supply chain:

= 350,000 tonnes/yr

According to Weiss et al. (16% of NF3 consumption X 17,200)

Every Hour, Linde prevents **40 tonnes** of CO2 equivalent emission reduction.

Clean Gas	Atmospheric Lifetime (Years)	Global Warming Potential (GWP <sub>100</sub> )
CF <sub>4</sub>	50,000	6,500
C <sub>2</sub> F <sub>6</sub>	10,000	9,200
C <sub>3</sub> F <sub>8</sub>	2,600	7,800
SF <sub>6</sub>	3,200	23,900
NF <sub>3</sub>	550	17,200
C <sub>5</sub> F <sub>8</sub>	0.98	90
COF <sub>2</sub>	1	1
C <sub>3</sub> F <sub>6</sub>	<<1	<1
F <sub>2</sub>	0	0

# Solution: Fluorine Chamber Cleaning



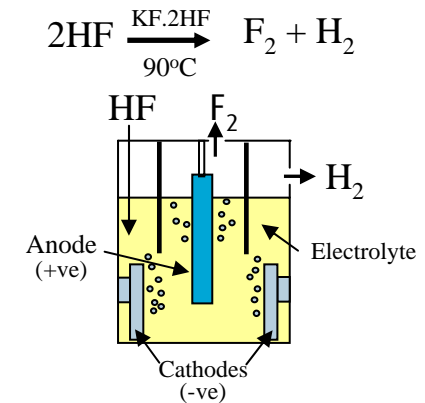
WHAT is Fluorine?

Fluorine is the active ingredient in all silicon chamber cleaning.

Fluorine gas (F<sub>2</sub>) is a faster and environmentally friendly replacement for Greenhouse Gases like NF<sub>3</sub> used to clean waste silicon from *Semicon, LCD and Solar CVD tools*

HOW do we deliver it?

Generation-F®  
On-site Fluorine  
Generation  
Systems



WHY use Fluorine?

- ✓ **Productivity** Faster cleaning boosts output **7-12%**, worth **€1.5/yr** to a typical Thin film solar customer
- ✓ **Cost Reduction** Direct material cost saving of **>20%**
- ✓ **Environment** F<sub>2</sub> has **Zero Global Warming Potential**, while NF<sub>3</sub> is **17.200 x CO<sub>2</sub>**

**Solution: On-site Fluorine,**  
模组化设计, 可依据用量而扩充



300 mm Wafer  
半导体十二寸晶圆厂



Gen 4 TFT-LCD



Gen 6 TFT-LCD



Gen 8 TFT-LCD  
平面显示器



1 tonne/yr



>10 tonnes/yr



100's tonnes/yr

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# Cleaning Performance = Productivity

氟气在制程上的好处=高产出(高处理量)

# On-site F<sub>2</sub> for Semiconductor dry cleaning: Process Experience since 2002



## Thermal clean for 300mm LPCVD

在半导体的主要应用是批次式LPCVD制程设备的干式清洁(热分解)

Improved selectivity demonstrated compared to ClF<sub>3</sub>

Improved etch rate (shorter cleaning time) compared to NF<sub>3</sub>

Reduced downtime compared to wet cleaning.

No restrictions in clean time due to cylinder size

100% F<sub>2</sub> available for cleaning – blending option available to meet any process requirement

Higher purity - potential benefits for film quality and / or clean frequency

→ Currently used in high volume 300mm production in : Korea, China, Singapore, France

→ OEM qualified and approved

## Movement to extend on-site F<sub>2</sub> to PECVD chamber cleaning in semiconductor

将氟气产生器的应用扩展到半导体PECVD的腔体清洁...

# Fluorine Process Benefits in PECVD Cleaning: A Combination of Thermodynamics (Energy) and Kinetics (Chemical Pathways)

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## Thermodynamics

F<sub>2</sub> takes much less energy to produce F· radicals than NF<sub>3</sub> or SF<sub>6</sub>

### Bond Energies [kJ / mol]

F <sub>2</sub>		NF <sub>3</sub>		SF <sub>6</sub>	
F---F	159	F2N---F	248	F5S---F	326
		FN---F	278	F4S---F	222
		N---F	316	O---O	249

- ➔ F· from F<sub>2</sub>: 60 kJ / mol
- ➔ F· from NF<sub>3</sub>: 281 kJ / mol
- ➔ F· from SF<sub>6</sub>: 399 kJ / mol

## Kinetics (in-situ)

F<sub>2</sub> has only one pathway in a plasma dissociation



It is kinetically forbidden for two atoms, like 2 F·, to recombine in the gas-phase

NF<sub>3</sub> has two competing pathways

1. NF<sub>3</sub> → NF<sub>2</sub> + F·
- 2a. NF<sub>2</sub>· → NF + F·
- 2b. NF<sub>2</sub>· + NF<sub>2</sub>· → N<sub>2</sub>F<sub>4</sub>

As process conditions tend to higher flows and rf power, pathway 2b predominates

O<sub>2</sub> is added to SF<sub>6</sub> plasmas to prevent sulfur deposits



The predominant pathway greatly limits efficiency

# High Performance PECVD Chamber cleaning

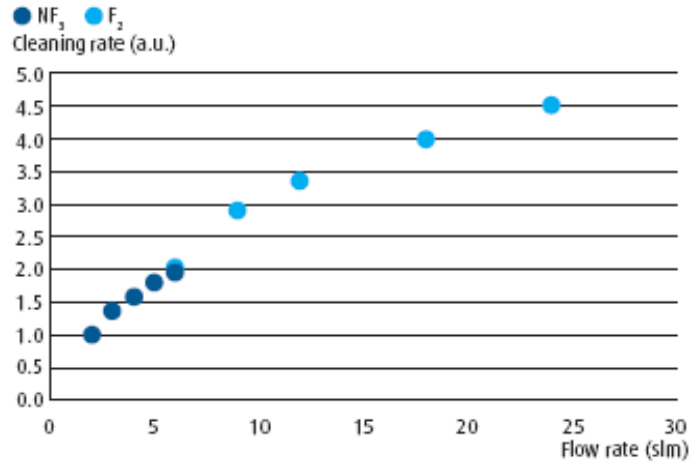
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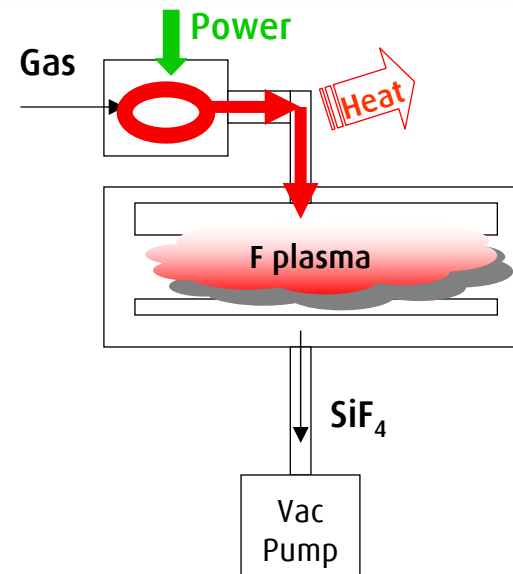
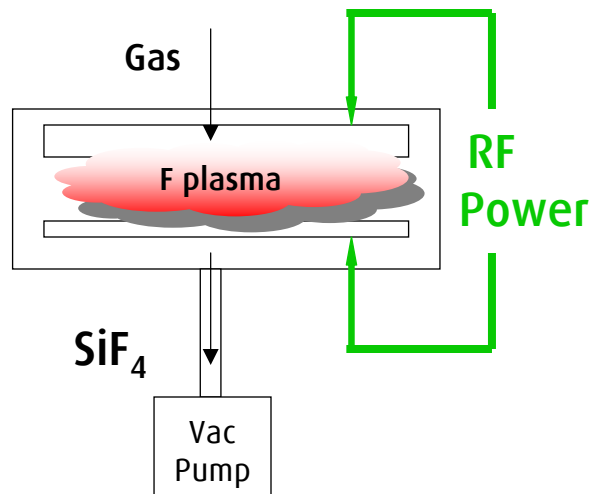
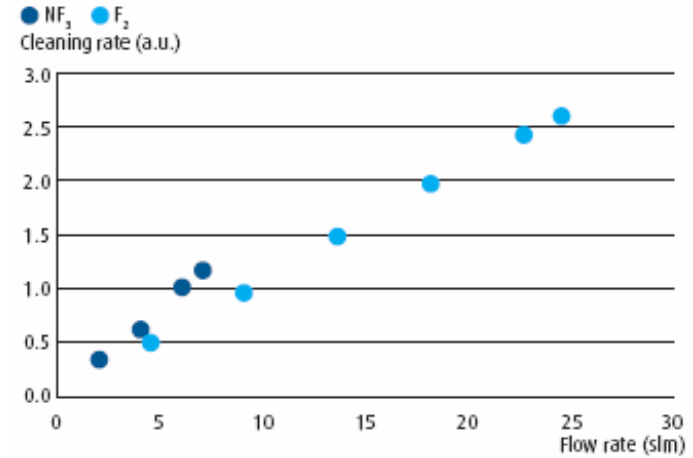
## In-situ Plasma

## Remote Plasma

F<sub>2</sub> vs NF<sub>3</sub> Chamber Cleaning



F<sub>2</sub> vs NF<sub>3</sub> Chamber Cleaning



# Fluorine Process Benefits: Cleaning Speed in PECVD: Summary of Actual Process Results



## RPS Activation

	F2	NF3
Mass Efficiency <sup>a</sup>	~100 %	80%
Cleaning Speed <sup>b</sup>	3x +	1x
Energy Consumption <sup>c</sup>	985 kWh / ton	2640 kWh / ton

## In Situ Activation

	F2	NF3	SF6
Mass Efficiency <sup>d</sup>	~100 %	55%	26%
Cleaning Speed <sup>e</sup>	5x +	1.7x	1x
Energy Consumption <sup>f</sup>	3280 kWh / ton	9580 kWh / ton	7110 kWh / ton

\*source:

- a) Malibu/Linde F2 tests at Bielefeld (AKT)
- b) Malibu/Linde F2 tests at Bielefeld (AKT), Linde F2 tests at ITRI (AKT)
- c) Malibu/Linde F2 tests at Bielefeld (AKT), Malibu/Linde F2 tests at Bielefeld (KAI)
- d) Linde F2 tests at NexPower (Ulvac) / Linde F2 test at Bielefeld (KAI)
- e) Oerlikon presentation at PVSEC09, Linde F2 tests at NexPower (Ulvac), Malibu/Linde F2 tests at Bielefeld (KAI)
- f) Linde F2 tests at NexPower (Ulvac), Malibu/Linde F2 tests at Bielefeld (KAI)



## 3<sup>rd</sup> Party Verification of Cleaning Performance:



- Publications on F2 Cleaning Speed, etch rate etc.

- Display:

- AKT Display: Published testing results FPD International, Yokohama, 2008

- Solar:

- AIST Japan: Published in-situ clean test results for solar
- Malibu PV published F2 performance results from their trials on Applied solar equipment

- Semiconductor:

- Fraunhofer Institute – etch rate studies
- Solvay Fluor, Infineon – etch rate studies

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# Cost Reduction

降低成本

# Fluorine Process Benefits: Mass saving



## RPS Activation

	F2	NF3
Mass Efficiency <sup>a</sup>	~100 %	80%

## In Situ Activation

	F2	NF3	SF6
Mass Efficiency <sup>b</sup>	~100 %	55%	26%

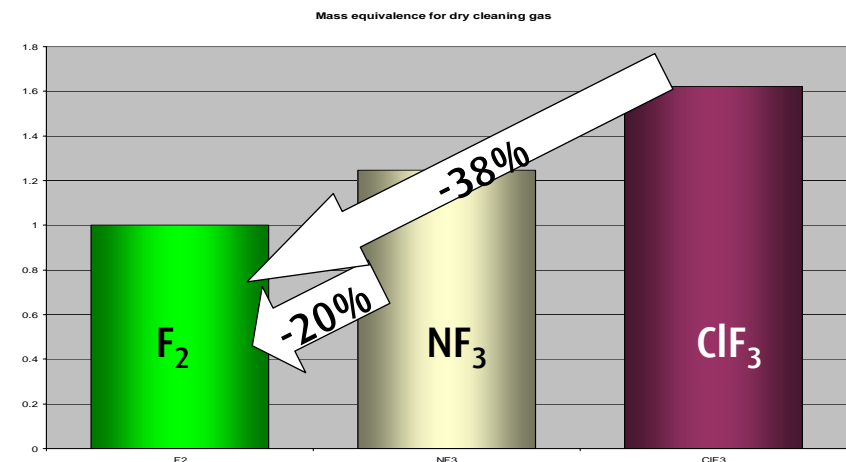
1.5 moles F2 required to produce same amount free Fluorine as 1 mole of NF3 or ClF3

50% higher flow F2 required compared to NF3 to provide same number of Fluorine radicals

**BUT:** 气体销售的单位不是以体积而是以重量

\*source:

- a) Malibu/Linde F2 tests at Bielefeld (AKT)
- b) Linde F2 tests at NexPower (Ulvac) / Linde F2 test at Bielefeld (KAI)



在完全解离的情况下, 1Kg F2, 1.25Kg NF3, 与 1.62Kg ClF3所能提供的氟自由基数量是相同的

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# CONCLUSION

结论

## Replacing $\text{NF}_3$ with On-site fluorine - addresses key throughput, cost & environmental concerns

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### Faster cleaning, higher productivity

- Fluorine can clean twice as fast as  $\text{NF}_3$  with 10% lower mass consumption
- Can increase line throughput from 3.5-10%

### Reduced gas consumption

- 80Kg of  $\text{F}_2$  delivers the same cleaning as 100 Kg of  $\text{NF}_3$  with the same clean rate

### Lower environmental impact

- $\text{F}_2$  has **zero** global warming potential vs. 17,000 for  $\text{NF}_3$
- 60% Reduced electricity consumption in plasma source

**Modular On Site Solution:  
Eliminates logistics of trailer supply chain and guarantees security of supply**

# CONCLUSION



Thanks for your attention,

Time for your questions / 提问时间...

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